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Investigation of the mechanism of Pt₃Fe₃ clusters for the hydrogen evolution reaction and for the oxygen reduction reaction

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With the fast-growing developing economies and the acceleration of industrial development, the demand for renewable, sustainable, and environmental-friendly energy sources is severely increasing [1]. Hydrogen, as a green fuel, attracts our attention and Pt stands out for its good performance in electrochemical production of hydrogen and for its use in fuel cells. However, the price of Pt limits its application and reducing the amount of Pt used in fabricating the electrodes becomes an urgent problem. According to our previous work [2], "Pt-based" clusters are an interesting option to reduce the loading of Pt and they are very stable thanks to a strong bond with the carbon support. Thence, we are studying Pt-X clusters (where X is a second metal) to further reduce the loading and to improve the activity. Therefore, we use Pt-X clusters as the electrode material and deposited it on the *rotating ring-disk electrode* (RRDE) tip to test the performance for the hydrogen evolution reaction (HER) and for the oxygen reduction reaction (ORR), in both alkaline and acidic media. We explored cyclic voltammetry curves (CV) by controlling the rotating speed of RRDE and the scan rate, using different working potential windows as well as N₂ and O₂ condition, in basic and acid media, to illustrate both reactions. Besides, we adopted X-ray absorption spectroscopies (XAS) to investigate the oxidation state and local structure changes of Pt and Fe under operative conditions. From above, we want to study structure-activity relations and provide rational design instructions related to Pt-based clusters and, more in general, to the electrode material for both the ORR and the HER.

[1] H. Lee, X. Wu, L. Sun, *Nanoscale*. 2020, 12, 4187-4218.

[2] M. Fracchia, P. Ghigna, M. Marelli, M. Scavini, A. Vertova, S. Rondinini, R. Della Pergola, A. Minguzzi. *New J. Chem.*, Accept